

(19)



Europäisches Patentamt

European Patent Office

Office européen des brevets



(11)

EP 1 516 909 A2

(12)

## EUROPEAN PATENT APPLICATION

(43) Date of publication:  
23.03.2005 Bulletin 2005/12

(51) Int Cl.7: **C10L 1/22**, C10L 1/14,  
C10L 10/00, C10L 10/04

(21) Application number: 04254935.2

(22) Date of filing: 17.08.2004

(84) Designated Contracting States:  
**AT BE BG CH CY CZ DE DK EE ES FI FR GB GR**  
**HU IE IT LI LU MC NL PL PT RO SE SI SK TR**  
Designated Extension States:  
**AL HR LT LV MK**

(72) Inventors:  
• Bennett, Joshua J.  
Richmond Virginia 23233 (US)  
• Schwab, Scott D.  
Richmond Virginia 23233 (US)

(30) Priority: 18.09.2003 US 665907

(74) Representative: Colmer, Stephen Gary et al  
Mathys & Squire  
120 Holborn  
London EC1N 2SQ (GB)

(71) Applicant: Afton Chemical Corporation  
Richmond, VA 23219 (US)

(54) **Method of reducing amount of peroxides, reducing fuel sediments and enhancing fuel system elastomer durability, fuel stability and fuel color durability**

(57) A reduction in the formation and presence of peroxides in low sulfur diesel fuels is obtained through the combination of those fuels with an organic nitrate combustion improver. The reduction in the amount of

peroxides means that the fuel system elastomers will be more durable, as they are not being corroded by as much peroxide formed in the fuel, fuel color durability is improved, fuel stability is enhanced, and fuel sediments are reduced.

## ULSD Test Fuels Selected Properties

Property	ASTM Method	Fuel ID	
		A	B
Cetane number	D 613	42.3	59.2
Aromatics, vol%	D 1319	24.8	21.7
Saturates, vol%	D 1319	73.6	76.3
Olefins, vol%	D 1319	1.6	2.0
Sulfur, ppm (wt.)	D 5453	1	17
Specific Gravity	D 4052	.831	.836
10% B.P., °C	D 86	213	257
50% B.P., °C	D 86	262	296
90% B.P., °C	D 86	322	338
Kinematic visc., 40°C, cSt	D 445	2.48	4.02

EP 1 516 909 A2

## Description

[0001] This present invention relates to a method including the addition of an organic nitrate combustion improver to a middle distillate fuel to reduce formation or presence in the fuel of peroxides. Especially in low or ultra-low sulfur fuels, the addition of an organic nitrate combustion improver, for instance 2-ethylhexyl nitrate, retards the formation and/or reduces the presence of peroxides, and prolongs the life of gaskets, hoses, seals and other elastomeric parts exposed to the peroxides. Other benefits include a reduction in fuel sediments, and enhanced fuel stability and color durability.

## Background

[0002] There is a current trend towards the use of ultra low sulfur diesel fuels, commonly referred to as fuels having 50 ppm sulfur or less ("ULSD fuels"). This trend toward the use of ULSD fuels has caused substantial combustion system changes and equally significant changes in fuel specifications. Many industrialized nations are reducing and/or have already reduced their mandatory maximum specifications for sulfur content. As a result, there are new concerns with respect to the performance and handling of the fuels formulated to meet the new specifications.

[0003] One concern with ULSD fuels is that the removal of sulfur compounds, some of which are effective peroxide scavengers and/or decomposers, may allow peroxides to build up in these fuels. The potential increase in peroxides is detrimental to fuel systems, because peroxides are known to degrade fuel system elastomers. The increase in peroxides, therefore, could cause the possible failure of seals, gaskets and hoses in a fuel system that uses ULSD fuels. See, for instance, Owen and Coley, *Automotive Fuels Reference Book, Second Edition*, 1995, pp. 520-523. The potential seriousness of this problem is also well documented in the problems with jet fuels in the 1960's and 1970's where high peroxide content in those fuels was associated with a high failure rate for fuel hoses, gaskets and seals in those systems. E.g., Fodor, et al., "Peroxide Formation in Jet Fuels," *Energy and Fuels*, 1988, pp. 729-34.

[0004] Other concerns that arise when peroxide levels increase include fuel stability, color durability, and fuel sediments. These concerns are discussed generally in Bacha and Lesnini, "Diesel Fuel Thermal Stability at 300°F," Sixth International Conference on Stability and Handling of Liquid Fuels, Vancouver, Canada, October 13-17, 1997; Vardi and Kraus, "Peroxide Formation in Low Sulfur Automotive Diesel Fuels," SAE Paper No. 920826.

[0005] It has been observed that combustion improvers may in fact promote the formation of peroxides at relatively higher temperatures. This observation is assumed true for all temperatures. Accordingly, there is a possible concern that ULSD fuels, and particularly those

containing combustion improvers, may have a propensity to form detrimental levels of peroxides and hydroperoxides during storage.

## Brief Description of the Drawings

### [0006]

Figure 1 is a chart characterizing the fuels that were tested as described herein.

Figure 2 is a graph demonstrating hydroperoxide kinetics of the fuels tested as described herein.

## Detailed Description

[0007] A reduction in the formation or presence of peroxides and hydroperoxides in ultra low sulfur diesel fuels is obtained through the combination of an organic nitrate combustion improver with the fuel. By reducing the amount and/or formation of peroxides and hydroperoxides, it is possible to enhance the durability of middle distillate fuel system elastomers, enhance fuel stability, enhance color durability and reduce formation of fuel sediments.

[0008] It is believed that the interaction between organic nitrate combustion improvers and peroxides/hydroperoxides includes a mechanism that is dependant on temperature. "Peroxides" is meant herein to include peroxides, hydroperoxides, mixtures thereof and precursors thereof. As demonstrated in testing discussed herein, fuels containing organic nitrate combustion improvers actually have increased peroxide levels over time as compared to fuels without any organic nitrate combustion improver, but only at high temperatures (temperatures greater than about 70°C). This finding generally of a higher amount of peroxides resulting from the use of organic nitrate combustion improvers is consistent with conventional wisdom. However, it has been discovered that at temperatures below about 70°C, there is actually an unexpected reduction in the amount or the formation of peroxides when an organic nitrate combustion improver is combined with an ULSD fuel.

[0009] A method of reducing the amount of peroxides in low sulfur, middle distillate fuels comprises the steps of: providing a middle distillate fuel having a sulfur content of about 50 ppm or less; combining the fuel with an organic nitrate combustion improver; wherein the amount of organic nitrate combustion improver combined with the fuel reduces the amount of peroxides in the fuel as compared with a middle distillate fuel without the organic nitrate combustion improver.

[0010] Fuels are rarely stored at temperatures of about 70°C or higher. If a fuel ever reaches that temperature in the operation of a combustion system, then the fuel would only remain at that temperature for a very short time before combustion. As a result, a relatively insignificant increase in peroxide presence and/or formation would result, if at all, from the use of an organic

nitrate combustion improver. More importantly, middle distillate fuel may often be stored for days/weeks/months before use. Typical storage temperatures would be well below 70°C. Realistically, therefore, it is significant that an organic nitrate combustion improver is discovered to retard the formation of or reduce the amount of peroxides in ULSD fuels.

**[0011]** A presentation entitled "Hydroperoxide Formation in Ultra-Low Sulfur Diesel Fuels" by Joshua J. Bennett and Scott D. Schwab was prepared for the International Conference on Stability and Handling of Liquid Fuels, Steamboat Springs, Colorado on September 19, 2003. That presentation and the materials presented are incorporated herein by reference as if set forth in their entirety.

**[0012]** The hydrocarbonaceous fuels utilized herein are comprised in general of mixtures of hydrocarbons which fall within the distillation range of about 160 to about 370° C. Such fuels are frequently referred to as "middle distillate fuels" since they comprise the fractions which distill after gasoline. Such fuels include diesel fuels, biodiesel and biodiesel-derived fuels, burner fuel, kerosenes, gas oils, jet fuels, and gas turbine engine fuels.

**[0013]** In an embodiment, applicable middle distillate fuels are those characterized by having the following distillation profile:

	°F	°C
IBP	250-500	121-260
10%	310-550	154-288
50%	350-600	177-316
90%	400-700	204-371
EP	450-750	232-399

**[0014]** Diesel fuels having a clear cetane number (i.e., a cetane number when devoid of any cetane improver such as an organic nitrate) in the range of 30 to 60 may also be used. In another example are those in which the clear cetane number is in the range of 40 to 50.

**[0015]** The organic nitrate combustion improvers (also frequently known as ignition improvers) comprise nitrate esters of substituted or unsubstituted aliphatic or cycloaliphatic alcohols which may be monohydric or polyhydric. The organic nitrates may be substituted or unsubstituted alkyl or cycloalkyl nitrates having up to about 10 carbon atoms, for example from 2 to 10 carbon atoms. The alkyl group may be either linear or branched (or a mixture of linear and branched alkyl groups). Specific examples of nitrate compounds suitable for use as nitrate combustion improvers include, but are not limited to the following: methyl nitrate, ethyl nitrate, n-propyl nitrate, isopropyl nitrate, allyl nitrate, n-butyl nitrate, isobutyl nitrate, sec-butyl nitrate, tert-butyl nitrate, n-amyl nitrate, isoamyl nitrate, 2-amyl nitrate, 3-amyl nitrate,

tert-amyl nitrate, n-hexyl nitrate, n-heptyl nitrate, sec-heptyl nitrate, n-octyl nitrate, 2-ethylhexyl nitrate, sec-octyl nitrate, n-nonyl nitrate, n-decyl nitrate, cyclopentyl nitrate, cyclohexyl nitrate, methylcyclohexyl nitrate, isopropylcyclohexyl nitrate, and the like. Also suitable are the nitrate esters of alkoxy substituted aliphatic alcohols such as 2-ethoxyethyl nitrate, 2-(2-ethoxyethoxy) ethyl nitrate, 1-methoxypropyl-2-nitrate, and 4-ethoxybutyl nitrate, as well as diol nitrates such as 1, 6-hexamethylene dinitrate and the like. For example the alkyl nitrates and dinitrates having from 5 to 10 carbon atoms, and most especially mixtures of primary amyl nitrates, mixtures of primary hexyl nitrates, and octyl nitrates such as 2-ethylhexyl nitrate are also included.

**[0016]** As is well known, nitrate esters are usually prepared by the mixed acid nitration of the appropriate alcohol or diol. Mixtures of nitric and sulfuric acids are generally used for this purpose. Another way to making nitrate esters involves reacting an alkyl or cycloalkyl halide with silver nitrate.

**[0017]** The concentration of nitrate ester or other organic nitrate combustion improver in the middle distillate fuel can be varied within relatively wide limits such that the amount employed is at least sufficient to cause a reduction in the presence and/or formation of peroxides. This amount may fall within the range of 100 to 5,000 parts by weight per million parts of fuel.

**[0018]** Other additives may be included within the fuel compositions described herein provided they do not adversely affect the amount or formation of peroxides otherwise obtained herein. Thus, use may be made of one or more of such components as corrosion inhibitors, antioxidants, anti-rust agents, detergents and dispersants, fuel lubricity additives, demulsifiers, dyes, inert diluents, cold flow improvers, conductivity agents, metal deactivators, stabilizers, antifoam additives, de-icers, biocides, odorants, drag reducers, combustion improvers, e.g., including MMT, oxygenates and like materials. These additives may also be used in combinations as additive packages.

**[0019]** Sulfur compounds themselves may reduce the amount of peroxide in a fuel, so the present analysis is directed to low sulfur fuels. For example, ultra-low sulfur fuels containing the organic nitrate combustion improver may have less than about 100 ppm sulfur, or alternatively, less than about 50 ppm sulfur. Still further alternatives include fuels having less than about 20 ppm or less than about 10 ppm of sulfur.

**[0020]** The advantages achievable from the addition of an organic nitrate combustion improver to a low sulfur fuel are demonstrated in the following tests. For purposes of these tests, it is deemed detrimental to have a concentration of peroxides greater than about 8 ppm. Accordingly, measurements made herein were with respect to time/temperature conditions of specific fuels which result in a concentration of a peroxide greater than about 8 ppm. First, two different fuels were tested. These fuels were identified as Fuel A and Fuel B. Fuels

having significantly different properties were identified in order to best evaluate how different fuels may have different results. Figure 1 defines the two Fuels A and B that were used in the testing.

[0021] Fuels A and B were each tested with and without the addition of 2500 ppm 2-ethylhexyl nitrate combustion improver. As a result of engine testing, it was determined the precise fuel conditions (temperature and residence time) which generate detrimental concentrations (greater than 8 ppm) of peroxides. Given the known points on the graph, lines were calculated to represent the peroxide formation kinetics. The specific peroxide kinetics that were indicated are shown in Figure 2.

[0022] As is evident from the kinetics demonstrated in Figure 2, the fuels containing the organic nitrate combustion improver (e.g., 2-EHN) each demonstrate a longer time to reach a detrimental level of peroxides when fuel temperatures are below approximately 70°C. The specific temperature at which such detrimental levels of peroxides would arise would be the intersection of the demonstrated linear kinetics for the fuels with and without the organic nitrate combustion improver.

[0023] The organic nitrate combustion improver with a middle distillate fuel enables each of the (1) elastomer durability benefit, (2) enhanced fuel stability, (3) fuel sediment reduction, and (4) enhanced color durability obtained by keeping the amount of peroxides in ULSD fuels less than about 8 ppm.

[0024] Based on the foregoing tests and calculation, it is seen that peroxide formation and/or presence (i.e., the amount of peroxide) is reduced in middle distillate fuels containing an organic nitrate combustion improver. This may be a significant benefit in prolonging the life of elastomeric materials contacting the fuels when the fuels are stored for long periods of time. Other benefits include enhanced fuel stability, color durability, and a reduction in fuel sediments.

[0025] It is expected that the durability of elastomers susceptible to degradation by exposure to peroxides in a fuel system might thus be extended or enhanced by at least 25% in terms of miles driven, gallons of fuel combusted or days/years of service as compared to the durability of elastomers in a fuel system not containing an organic nitrate combustion improver. In another embodiment, the elastomer durability is extended or enhanced by at least 10% as compared to the durability of elastomers exposed to fuels not containing an organic nitrate combustion improver.

[0026] It is expected that the fuel stability of a middle distillate fuel might thus be extended or enhanced by at least 25% in terms of miles driven, gallons of fuel combusted or days/years of service as compared to the fuel stability of a fuel not containing an organic nitrate combustion improver. In another embodiment, the fuel stability is extended or enhanced by at least 10% as compared to the stability of fuels not containing an organic nitrate combustion improver.

[0027] It is expected that the durability of fuel color

might thus be extended or enhanced by at least 25% in terms of miles driven, gallons of fuel combusted or days/years of service as compared to the durability of fuel color in a fuel not containing an organic nitrate combustion improver. In another embodiment, the fuel color durability is expected to be extended or enhanced by at least 10% as compared to the durability of fuels not containing an organic nitrate combustion improver.

[0028] It is expected that the formation or presence of fuel sediments is reduced by at least 25% in terms of miles driven, gallons of fuel combusted or days/years of service as compared to the amount of fuel sediments in a fuel not containing an organic nitrate combustion improver. In another embodiment, the amount of fuel sediments is reduced or enhanced by 10% as compared to the amount of fuel sediments in fuels not containing an organic nitrate combustion improver.

[0029] It is to be understood that the reactants and components referred to by chemical name anywhere in the specification or claims hereof, whether referred to in the singular or plural, are identified as they exist prior to coming into contact with another substance referred to by chemical name or chemical type (e.g., base fuel, solvent, etc.). It matters not what chemical changes, transformations and/or reactions, if any, take place in the resulting mixture or solution or reaction medium as such changes, transformations and/or reactions are the natural result of bringing the specified reactants and/or components together under the conditions called for pursuant to this disclosure. Thus the reactants and components are identified as ingredients to be brought together either in performing a desired chemical reaction or in forming a desired composition (such as an additive concentrate or additized fuel blend). It will also be recognized that the additive components can be added or blended into or with the base fuels individually per se and/or as components used in forming preformed additive combinations and/or subcombinations. Accordingly, even though the claims hereinafter may refer to substances, components and/or ingredients in the present tense ("comprises", "is", etc.), the reference is to the substance, components or ingredient as it existed at the time just before it was first blended or mixed with one or more other substances, components and/or ingredients in accordance with the present disclosure. The fact that the substance, components or ingredient may have lost its original identity through a chemical reaction or transformation during the course of such blending or mixing operations or immediately thereafter is thus wholly immaterial for an accurate understanding and appreciation of this disclosure and the claims thereof.

#### Claims

1. A method of reducing the amount of peroxides in low sulfur, middle distillate fuels comprising the steps of:

providing a middle distillate fuel having a sulfur content of about 50 ppm or less;  
combining the fuel with an organic nitrate combustion improver;

wherein the amount of organic nitrate combustion improver combined with the fuel reduces the amount of peroxides in the fuel as compared with a middle distillate fuel without the organic nitrate combustion improver.

2. A method of enhancing the durability of middle distillate fuel system elastomers comprising the steps of:

providing a middle distillate fuel having a sulfur content of about 50 ppm or less;  
combining the fuel with an organic nitrate combustion improver;

wherein the amount of organic nitrate combustion improver combined with the fuel enhances the durability of middle distillate fuel system elastomers as compared with the durability of elastomers in a middle distillate fuel system combusting a middle distillate fuel without the organic nitrate combustion improver.

3. A method according to Claim 2, wherein the durability of the elastomers is enhanced by up to 25% as measured by miles driven, gallons of fuel combusted or days/years of service, relative to the durability of elastomers in a middle distillate fuel system combusting fuel without an organic nitrate combustion improver.

4. A method of enhancing the color durability of a middle distillate fuel comprising the steps of:

providing a middle distillate fuel having a sulfur content of about 50 ppm or less;  
combining the fuel with an organic nitrate combustion improver;

wherein the amount of organic nitrate combustion improver combined with the fuel enhances the color durability of said middle distillate fuel compared with the color durability of a middle distillate fuel without the organic nitrate combustion improver.

5. A method according to Claim 4, wherein the fuel color durability is enhanced by up to 25% as measured by miles driven, gallons of fuel combusted or days/years of service, relative to the color durability of fuels without an organic nitrate combustion improver.

6. A method of enhancing the fuel stability of a middle distillate fuel comprising the steps of:

providing a middle distillate fuel having a sulfur content of about 50 ppm or less;  
combining the fuel with an organic nitrate combustion improver;

wherein the amount of organic nitrate combustion improver combined with the fuel enhances the fuel stability of said middle distillate fuel as compared with the fuel stability of a middle distillate fuel without the organic nitrate combustion improver.

7. A method according to Claim 6, wherein the fuel stability enhanced by up to 25% as measured by miles driven, gallons of fuel combusted or days/years of service, relative to the fuel stability of fuels without an organic nitrate combustion improver.

8. A method of reducing fuel sediment in a middle distillate fuel comprising the steps of:

providing a middle distillate fuel having a sulfur content of about 50 ppm or less;  
combining the fuel with an organic nitrate combustion improver;

wherein the amount of organic nitrate combustion improver combined with the fuel reduces fuel sediments in the middle distillate fuel as compared with the fuel sediments in a middle distillate fuel without the organic nitrate combustion improver.

9. A method according to Claim 8, wherein the fuel sediment is reduced by up to 25% as measured by miles driven, gallons of fuel combusted or days/years of service, relative to the fuel sediment in a fuel without an organic nitrate combustion improver.

10. A method according to any one of the preceding claims wherein the organic nitrate combustion improver comprises a nitrate ester of a substituted or unsubstituted, aliphatic or cycloaliphatic, monohydric or polyhydric alcohol.

11. A method according to Claim 10 wherein the nitrate ester comprises 2-ethylhexyl nitrate.

12. A method according to any one of the preceding claims wherein the nitrate ester is combined in an amount from 100 to 5000 ppm wt. of the fuel, for example in an amount of about 2500 ppm wt. of the fuel.

13. A method according to any one of the preceding claims wherein the middle distillate fuel is selected

from the group consisting of diesel fuel, biodiesel fuel, burner fuel, kerosene, gas oil, jet fuel, and gas turbine engine fuel.

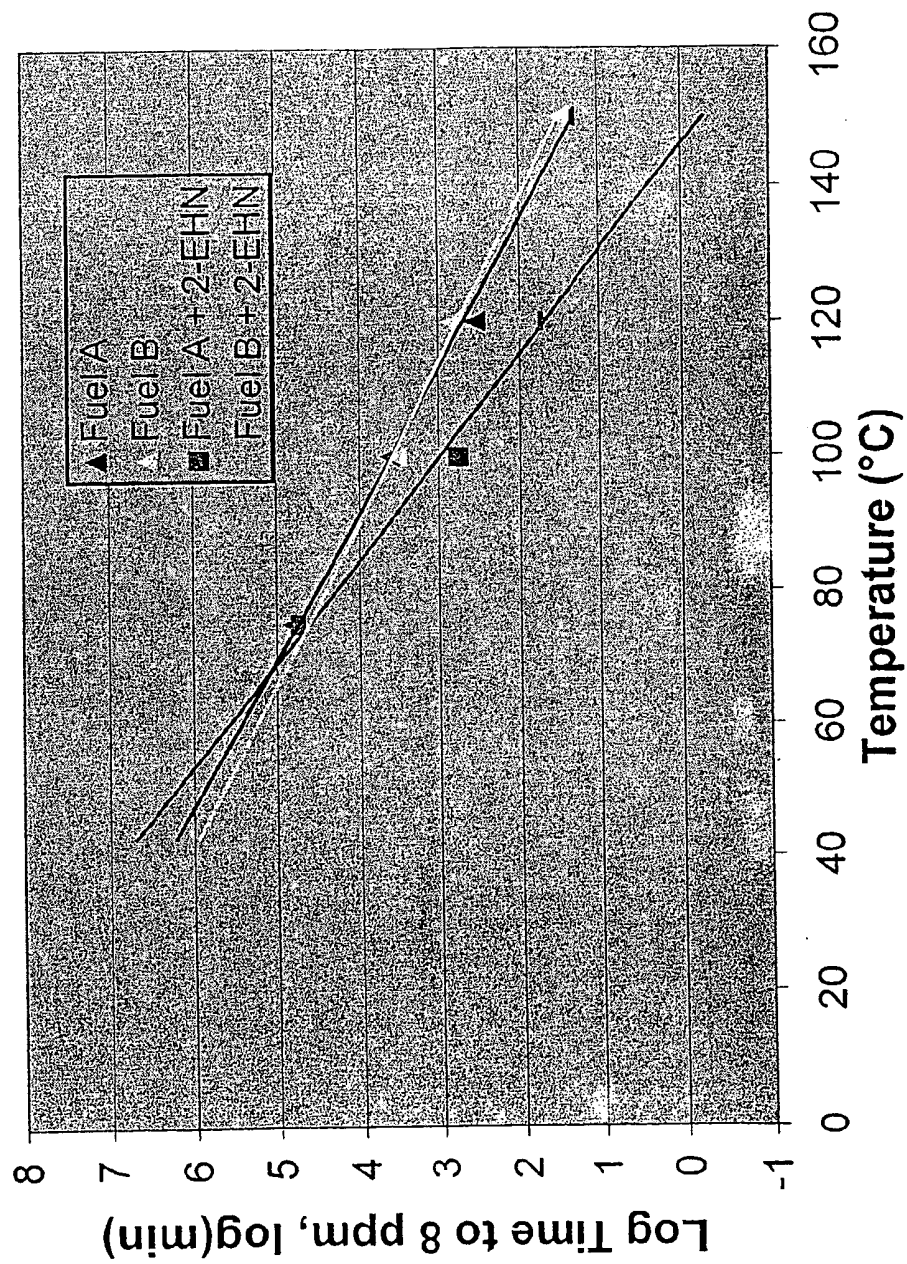
14. A method according to any one of the preceding claims wherein the fuel has a sulfur content of about 20 ppm or less, for example of about 10 ppm or less. 5
15. A method according to any one of the preceding claims wherein the fuel further comprises one or more components selected from the group consisting of: corrosion inhibitors, antioxidants, anti-rust agents, detergents and dispersants, fuel lubricity additives, demulsifiers, dyes, inert diluents, cold flow improvers, conductivity agents, metal deactivators, stabilizers, antifoam additives, de-icers, biocides, odorants, drag reducers, combustion improvers, MMT, and oxygenates. 10 15
16. A method according to any one of the preceding claims wherein the amount of peroxides in the fuel is less than about 8 ppm. 20
17. Use in a low sulfur, middle distillate fuel of an organic nitrate combustion improver in an amount sufficient to produce an amount of peroxides therein of less than about 8 ppm. 25
18. Use according to Claim 17 which results in improved elastomer durability in elastomeric components which contact the fuel. 30
19. Use according to Claim 17 or 18 which results in improved color durability of the fuel. 35
20. Use according to Claim 17, 18 or 19 which results in improved fuel stability. 40
21. Use according to any one of Claims 17 to 20 which results in reduction in fuel sediment. 45
22. Use according to any one of Claims 17 to 21 wherein the organic nitrate combustion improver comprises a nitrate ester of a substituted or unsubstituted, aliphatic or cycloaliphatic, monohydric or polyhydric alcohol. 50
23. Use according to any one of Claims 17 to 22 wherein the nitrate ester is combined in an amount from 100 to 5000 ppm wt. of the fuel, for example in an amount of about 2500 ppm wt. of the fuel. 55

# ULSD Test Fuels

## Selected Properties

Property	ASTM Method	Fuel ID	
		A	B
Cetane number	D 613	42.3	59.2
Aromatics, vol%	D 1319	24.8	21.7
Saturates, vol%	D 1319	73.6	76.3
Olefins, vol%	D 1319	1.6	2.0
Sulfur, ppm (wt.)	D 5453	1	17
Specific Gravity	D 4052	.831	.836
10% B.P., °C	D 86	213	257
50% B.P., °C	D 86	262	296
90% B.P., °C	D 86	322	338
Kinematic visc., 40°C, cSt	D 445	2.48	4.02

# Hydroperoxide Kinetics





(19)



Europäisches Patentamt

European Patent Office

Office européen des brevets



(11)

EP 1 516 909 A3

(12)

## EUROPEAN PATENT APPLICATION

(88) Date of publication A3:  
22.06.2005 Bulletin 2005/25

(51) Int Cl.7: **C10L 1/22**, C10L 1/14,  
C10L 10/00, C10L 10/04,  
C10G 75/02

(43) Date of publication A2:  
23.03.2005 Bulletin 2005/12

(21) Application number: 04254935.2

(22) Date of filing: 17.08.2004

(84) Designated Contracting States:  
**AT BE BG CH CY CZ DE DK EE ES FI FR GB GR**  
**HU IE IT LI LU MC NL PL PT RO SE SI SK TR**  
Designated Extension States:  
**AL HR LT LV MK**

(72) Inventors:  
• **Bennett, Joshua J.**  
Richmond Virginia 23233 (US)  
• **Schwab, Scott D.**  
Richmond Virginia 23233 (US)

(30) Priority: 18.09.2003 US 665907

(74) Representative: **Colmer, Stephen Gary et al**  
**Mathys & Squire**  
120 Holborn  
London EC1N 2SQ (GB)

(71) Applicant: **Afton Chemical Corporation**  
Richmond, VA 23218-2158 (US)

(54) **Method of reducing amount of peroxides, reducing fuel sediments and enhancing fuel system elastomer durability, fuel stability and fuel color durability**

(57) A reduction in the formation and presence of peroxides in low sulfur diesel fuels is obtained through the combination of those fuels with an organic nitrate combustion improver. The reduction in the amount of

peroxides means that the fuel system elastomers will be more durable, as they are not being corroded by as much peroxide formed in the fuel, fuel color durability is improved, fuel stability is enhanced, and fuel sediments are reduced.

## ULSD Test Fuels Selected Properties

Property	ASTM Method	Fuel ID	
		A	B
Cetane number	D 613	42.3	59.2
Aromatics, vol%	D 1319	24.8	21.7
Saturates, vol%	D 1319	73.6	76.3
Olefins, vol%	D 1319	1.6	2.0
Sulfur, ppm (wt.)	D 5453	1	17
Specific Gravity	D 4052	.831	.836
10% B.P., °C	D 86	213	257
50% B.P., °C	D 86	262	296
90% B.P., °C	D 86	322	338
Kinematic visc., 40°C, cSt	D 445	2.48	4.02

EP 1 516 909 A3



European Patent  
Office

# EUROPEAN SEARCH REPORT

Application Number  
EP 04 25 4935

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
X	EP 1 321 504 A (ETHYL CORPORATION) 25 June 2003 (2003-06-25) * page 3, line 34 - line 56; claims 1-14; tables 1,2 * * page 4, line 51 - line 57 * * page 6, line 30 - line 40 * -----	1-23	C10L1/22 C10L1/14 C10L10/00 C10L10/04 C10G75/02
A	US 3 239 320 A (JR. PAUL E. OBERDORFER,) 8 March 1966 (1966-03-08) * claims 1-5; examples 1-4 * -----	1,4,6,8	
A	US 2 051 873 A (JR. CHARLES D. LOWRY,) 25 August 1936 (1936-08-25) * column 1, line 39 - line 53; claims 1,5-7 * * column 2, line 55 - column 3, line 9 * -----	1,4,6,8	
A	GB 2 158 091 A (* BANKAMERICA CORPORATION) 6 November 1985 (1985-11-06) * claims 1-5,14-25,29 * -----	2	
A	US 2 158 050 A (BERESLAVAKY EUPHIME V) 16 May 1939 (1939-05-16) * column 4, line 65 - column 5, line 9; claims 7,8 * * column 6, line 55 - line 70 * -----	1,2,4,6, 8,10-13	TECHNICAL FIELDS SEARCHED (Int.Cl.7) C10L C10G
The present search report has been drawn up for all claims			
Place of search <b>The Hague</b>		Date of completion of the search <b>28 April 2005</b>	Examiner <b>de La Morinerie, B</b>
<b>CATEGORY OF CITED DOCUMENTS</b> X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	

4  
EPO FORM 1503 03/82 (P04C01)

# BEST AVAILABLE COPY

EP 1 516 909 A3

## ANNEX TO THE EUROPEAN SEARCH REPORT ON EUROPEAN PATENT APPLICATION NO.

EP 04 25 4935

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on  
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

28-04-2005

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP 1321504 A	25-06-2003	US 2003110684 A1	19-06-2003
		CA 2411933 A1	18-06-2003
		EP 1321504 A2	25-06-2003
		JP 2003193069 A	09-07-2003
US 3239320 A	08-03-1966	NONE	
US 2051873 A	25-08-1936	NONE	
GB 2158091 A	06-11-1985	AT 41916 T	15-04-1989
		AU 570106 B2	03-03-1988
		AU 4129685 A	24-10-1985
		BR 8501802 A	10-12-1985
		CA 1243047 A1	11-10-1988
		DE 3569234 D1	11-05-1989
		EP 0161705 A2	21-11-1985
		ES 8700646 A1	16-01-1987
		GB 2158090 A ,B	06-11-1985
		IN 164493 A1	25-03-1989
		JP 1911437 C	09-03-1995
		JP 6035395 B	11-05-1994
		JP 60233020 A	19-11-1985
		US 4579986 A	01-04-1986
		ZA 8502817 A	27-11-1985
		AU 579341 B2	24-11-1988
		AU 3812385 A	07-11-1985
		BR 8501963 A	24-12-1985
		CA 1257772 A1	25-07-1989
		CN 86100480 A	19-08-1987
		DE 3515567 A1	07-11-1985
		FR 2563839 A1	08-11-1985
		IL 75003 A	30-11-1988
		IN 163994 A1	24-12-1988
		JP 61000293 A	06-01-1986
		NO 851732 A ,B,	04-11-1985
		SE 460199 B	18-09-1989
		SE 8502093 A	02-11-1985
US 2158050 A	16-05-1939	NONE	

EPO FORM P4459

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82

**THIS PAGE BLANK (USPTO)**